

# **The Use of Coatings to Improve the Physical and Analytical Reliability of Process Monitors Used for Ammonia, Mercury and Hydrogen Sulfide Service**

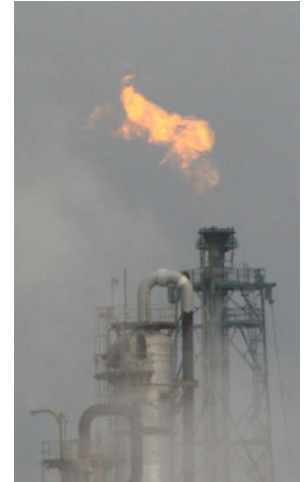
Gary Barone, Marty Higgins, David Smith  
SilcoTek Corporation

# Overview

- Sampling & Test Reliability
- Coating Materials
- Corrosion Testing
- Hydrophobicity comparison
- Chemical Inertness
- Conclusion



# Why Customers Need Reliability



- Reduce cost:
  - Lost time from re-testing
  - Improve product yields
  - Improve accuracy in grading feedstock
  - Emissions compliance
- Avoid false negatives and improve sample transfer
  - Sample stable from field to lab.
- Immediate response during process changes creates savings

# Factors Contributing to Poor Sampling Reliability

- Durability/Wear
- Corrosion
- Moisture
- Design

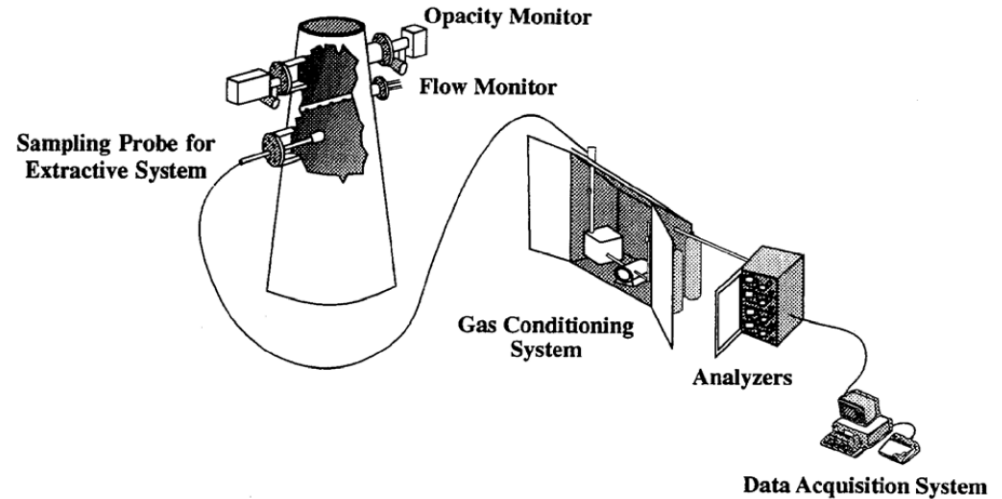


Figure 1: A typical continuous emission monitoring system (CEMS) (U.S. EPA Image)<sup>4</sup>

- Installation
- Chemical & Material Compatibility/Inertness
- Instrument Compatibility
- Inertness / Adsorption

# Using Coatings

- Most analytical pathways are stainless steel
  - Great structurally
  - Good corrosion resistance
  - Poor chemical properties for analytical chemists
- Coatings are now commonly used to improve the chemical properties.
- Need also to address corrosion and wear in harsh environments

# Selecting Coatings

- Fluoropolymers
  - Very inert
  - Very corrosion resistant
  - Broad pH applicability
  - Poor adhesion
  - Poor wear resistance
  - Good to 260°C
- Silicon based (Sulfinert<sup>®</sup>; SilcoNert<sup>™</sup>)
  - Very inert
  - Great adhesion
  - No carryover
  - Good corrosion resistance
  - Limited pH range
  - Susceptible to steam cleaning
  - Poor wear resistance
  - Good to 450°C

# New Coating

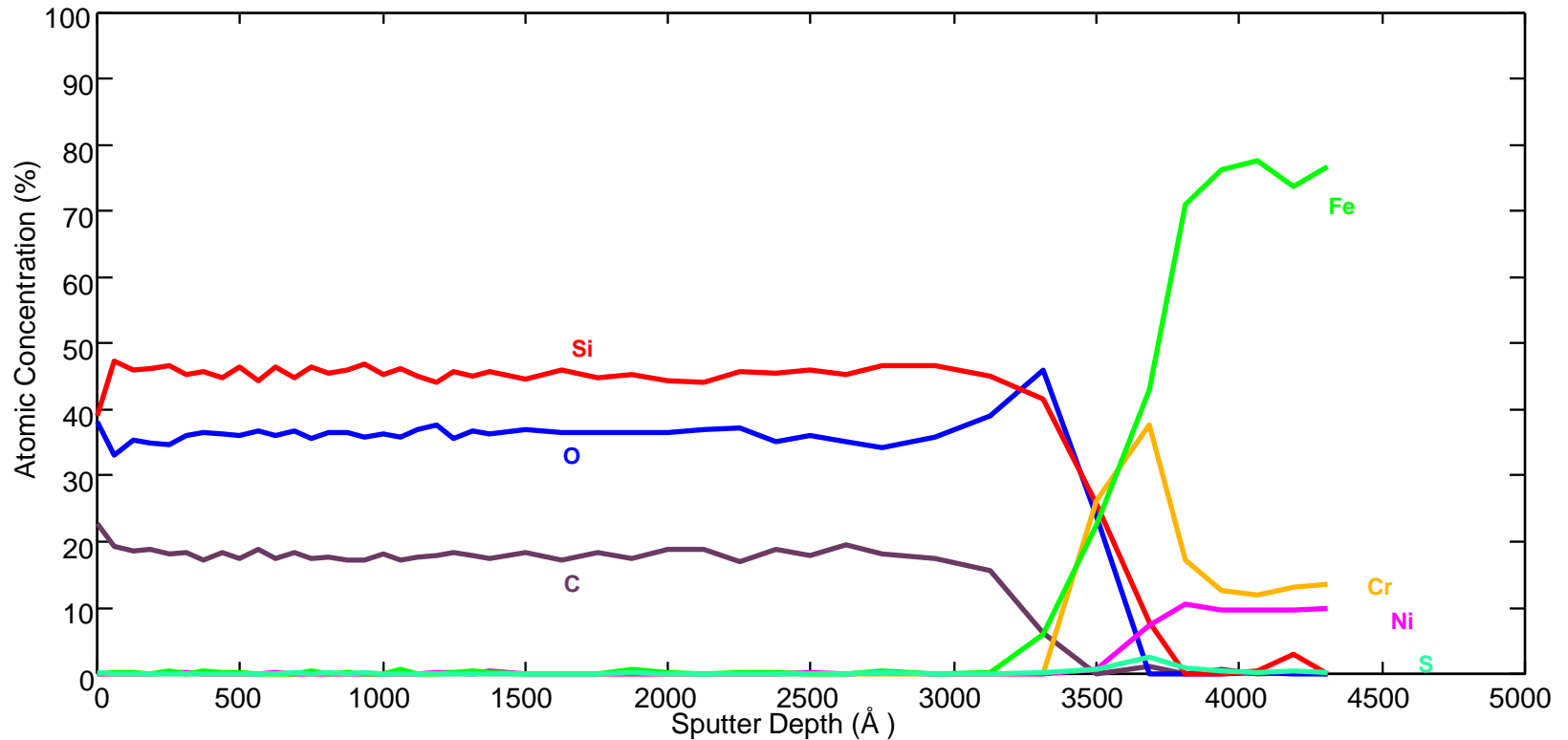
- Carboxysilane (Dursan™)
    - Good inertness
    - Great adhesion
    - No carryover
    - Good corrosion resistance
    - Broad pH applicability
    - Steam cleaning, no problem
    - Good wear resistance
    - Tested to 450°C so far
    - Still accumulating application data
-

# Coating/Material Properties

Property	SilcoNert 2000	Dursan	PTFE, PFA
Max Temperature	450°C	450°C	260°C
Min Temperature	-196°C	-100°C	-240°C
Low pH limit	0	0	0
High pH limit	7	14	14
Thickness	0.12um to 0.5um	0.5um to 1.5um	25um
Adhesion	Very Good	Very Good	Poor
Wear resistance	90% of Stainless	2 times 316 Stainless	10% of SS (est.)
Moisture contact	87°	104-140°	125°
Inertness vs. SS	Very Good	Very Good	Very Good



# Composition of Dursan™



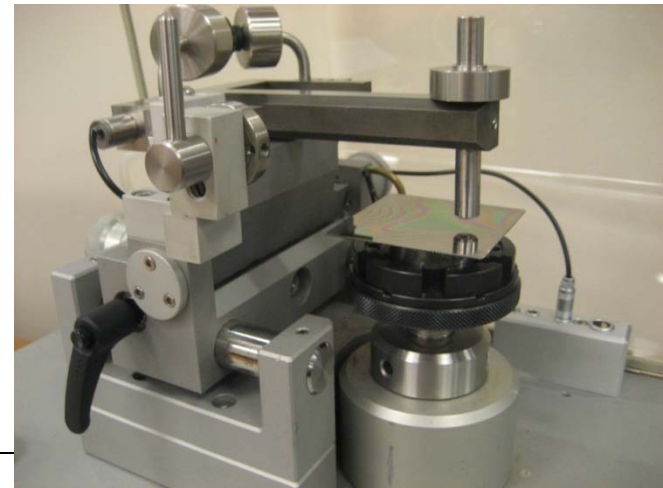
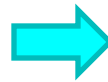
# Improving wear resistance

- Equipment and sample conditions can present physical challenges
  - Valve cycling
  - Particulate in sample streams
  - Cleaning needs
- Existing coatings that are applied to improve chemical inertness and control corrosion can be removed easily in challenging environments

# Wear Resistance Comparison

<i>Pin on Disc; 2.0N</i>	316 SS	Silco	Dursan
<b>Wear rate (x10<sup>-5</sup>mm<sup>3</sup>/N m)</b>	13.810	15.344	6.129
<b>Improvement Factor over Stainless Steel</b>	---	0.9 X	2 X

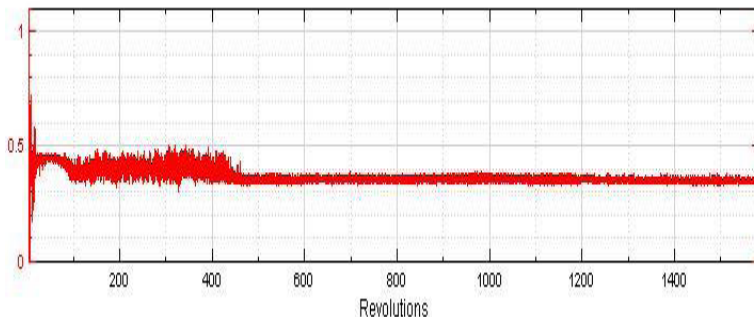
CSM Instruments  
Tribometer 18-343 used to  
measure surface wear  
resistance



# Wear and Friction Data

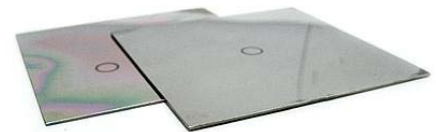
- **Pin on Disc in accordance with ASTM G133**

	<u>Avg. Coeff. Friction</u>	<u>Wear Rate</u> ( $\times 10^{-5} \text{mm}^3/\text{Nm}$ )
Uncoated SS	0.589	13.810
Carboxysilane on SS	0.378	6.129



**SilcoTek**  
Driving Innovation

<b>Load</b>	<b>2.0 N</b>
Duration	20 min
Speed	80 rpm
Radius	3mm
Revolutions	1,554
Ball Diameter	6mm
Ball Material	SS 440



Courtesy of Nanovea Inc.

# Challenge of Corrosion

- Samples can contain corrosives that quickly attack stainless
    - Hydrochloric acid (HCl)
    - Sulfuric acid (H<sub>2</sub>SO<sub>4</sub>)
    - Saltwater
  - Physical loss of equipment due to corrosion
    - Maintenance
    - Replacement cycles
  - Chemical inertness suffers with corrosion
  - Silicon coatings susceptible to caustics
-

# Comparative Corrosion Resistance

- *3M HCl; 24hr; 22°C*

<i>ASTM G31</i>	316L SS	Silco	Dursan
MPY	67.93	14.85	5.14
Improvement Factor over 316L stainless	---	4.6	13.2



# Acid Corrosion Resistance

- *ASTM G31 Guidelines: 6M HCl; 24hr; 23°C*

	316L SS	a-Silicon	Carboxysilane
MPY	181.98	4.32	0.44
Improvement Factor over 316L stainless	---	42	411

Photo after 19hr  
exposure



SiCO coated

Control / a-Si coated

# Comparative Corrosion Resistance

- *10% H<sub>2</sub>SO<sub>4</sub>; 24hr; 22°C*

<i>ASTM G31</i>	<b>316L SS</b>	<b>Silco</b>	<b>Dursan</b>
<b>MPY</b>	22.35	2.52	2.42
<b>Improvement Factor over 316L stainless</b>	---	8.9	9.9



# Exposure to Caustics

- *1M KOH; 24hr; 22°C*

<i>ASTM G31</i>	316L SS	Silco	Dursan
MPY	0	3.40	0.01
Improvement Factor Over Silicon	Infinite	--	261

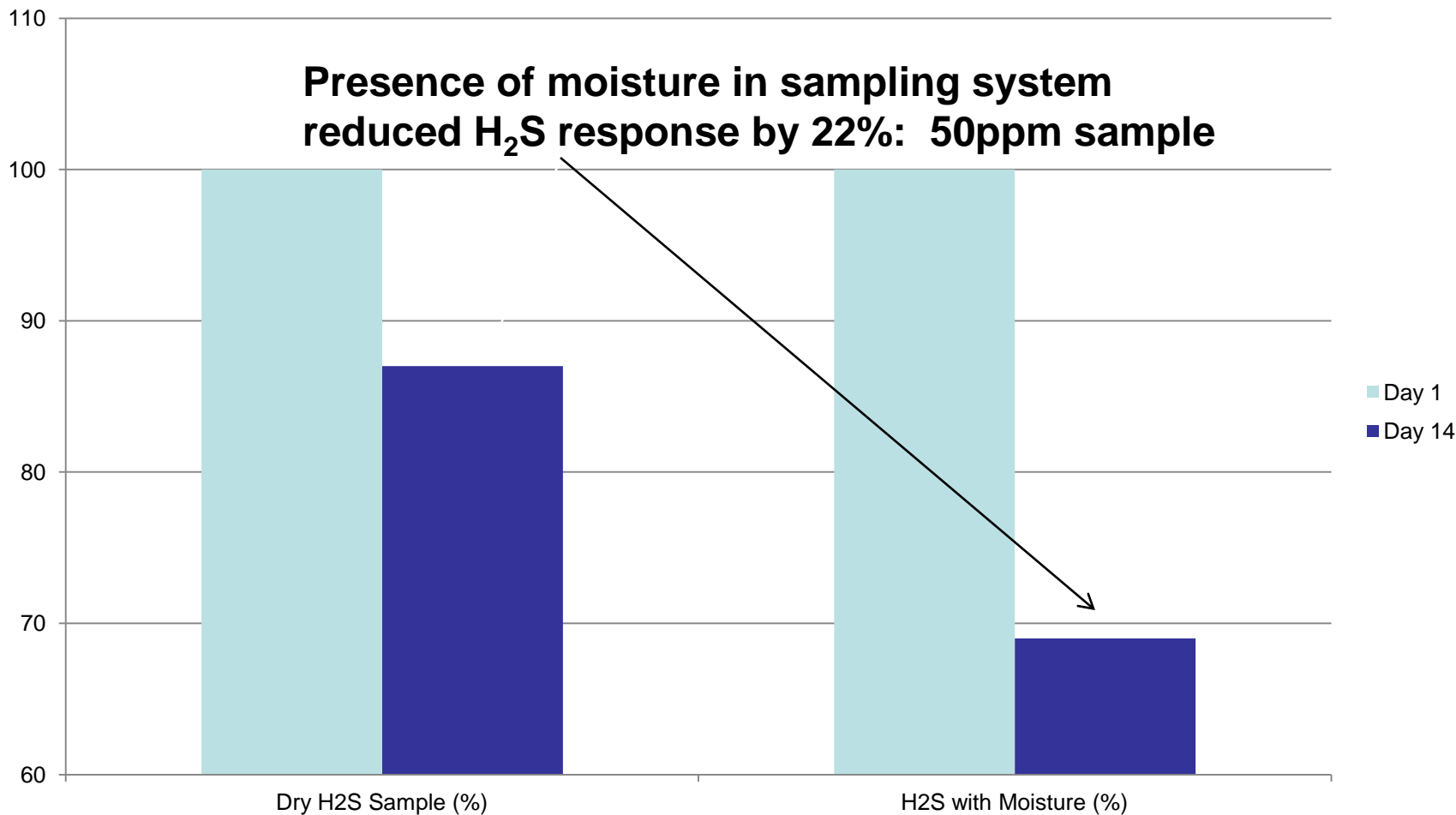
# Challenges of Moisture

- Adsorption of active compounds into entrained water
- Formation of acids within the sample system
- Formation of adsorptive rust particles in the sample system.
- Sticks to steel need parts to shed water

# Challenges of Moisture

- Benefits of coating that help release water faster
  - Components less susceptible to corrosion
  - Faster cycle times
  - Increased accuracy
  - Eliminate moisture/sample interaction

# Impact of Moisture

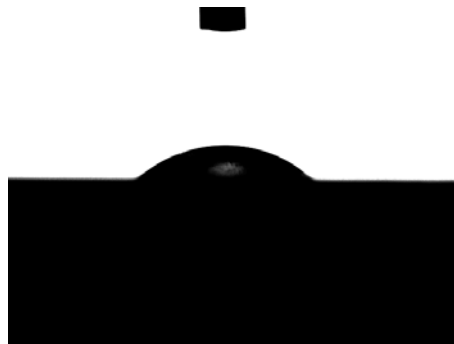


# Measuring Hydrophobicity

Kruss K100  
Tensiometer  
Testing on  
304 SS  
¼" OD tubing  
→

<i>DI Water</i>	<b>304 SS</b>	<b>Silicon</b>	<b>Sulfinert</b>	<b>Dursan</b>	<b>PTFE</b>
<b>Advancing</b>	36.0	53.6	87.3	105.5	125.4
<b>Receding</b>	5.3	19.6	51.5	85.3	84

DI Water Contact Angle Illustrations (advancing) on flat surfaces



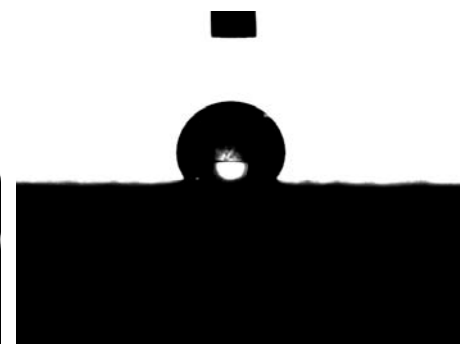
304 SS



Silicon



Sulfinert



Dursan

# Chemical Inertness

- Stainless Steel:
  - Adsorbs sulfur compounds
  - Causes loss of mercury
  - Demonstrates poor transportability (tailing) of polar organics such as alcohols
- Passivation when sulfur sampling shown effective at low temperatures, smooth surfaces<sup>1</sup>
  - Not effective with H<sub>2</sub>S in heated sample lines<sup>1</sup>
    - <sup>1</sup> Biela, B., Moore, R., Benesch, R., Talbert, B., Jacksier., “The Do’s and Don’ts in the Analysis of sulfur for Polyolefin Producers”, presented as paper 081 at the Gulf Coast Conference, (2003)

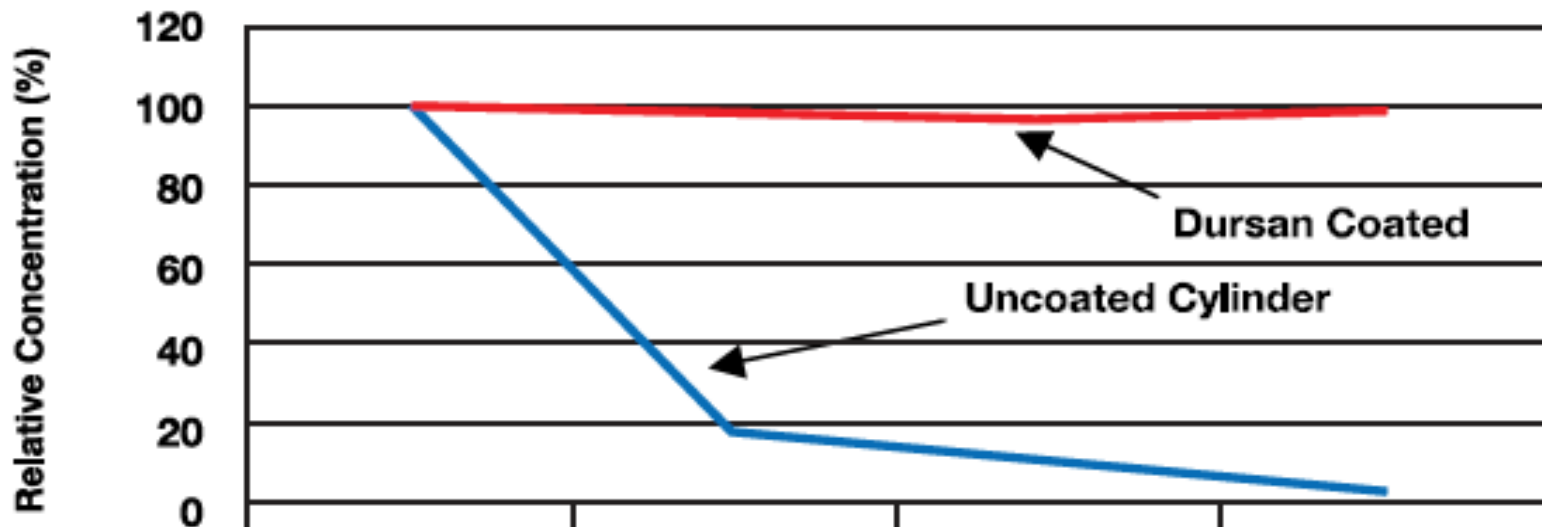
# Chemical Inertness

- SilcoNert™ and Fluoropolymers: Great inertness for sulfurs, mercury & ammonia, down to single digit ppb
  - Dursan™: Good for H<sub>2</sub>S (10ppm) and OK for Ammonia.
    - Improvements to target replacing Silco coatings for inertness and corrosive applications!
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# Hydrogen Sulfide

## H<sub>2</sub>S Stability: Dursan vs. Stainless Steel

50ppmv, 300cc cylinder

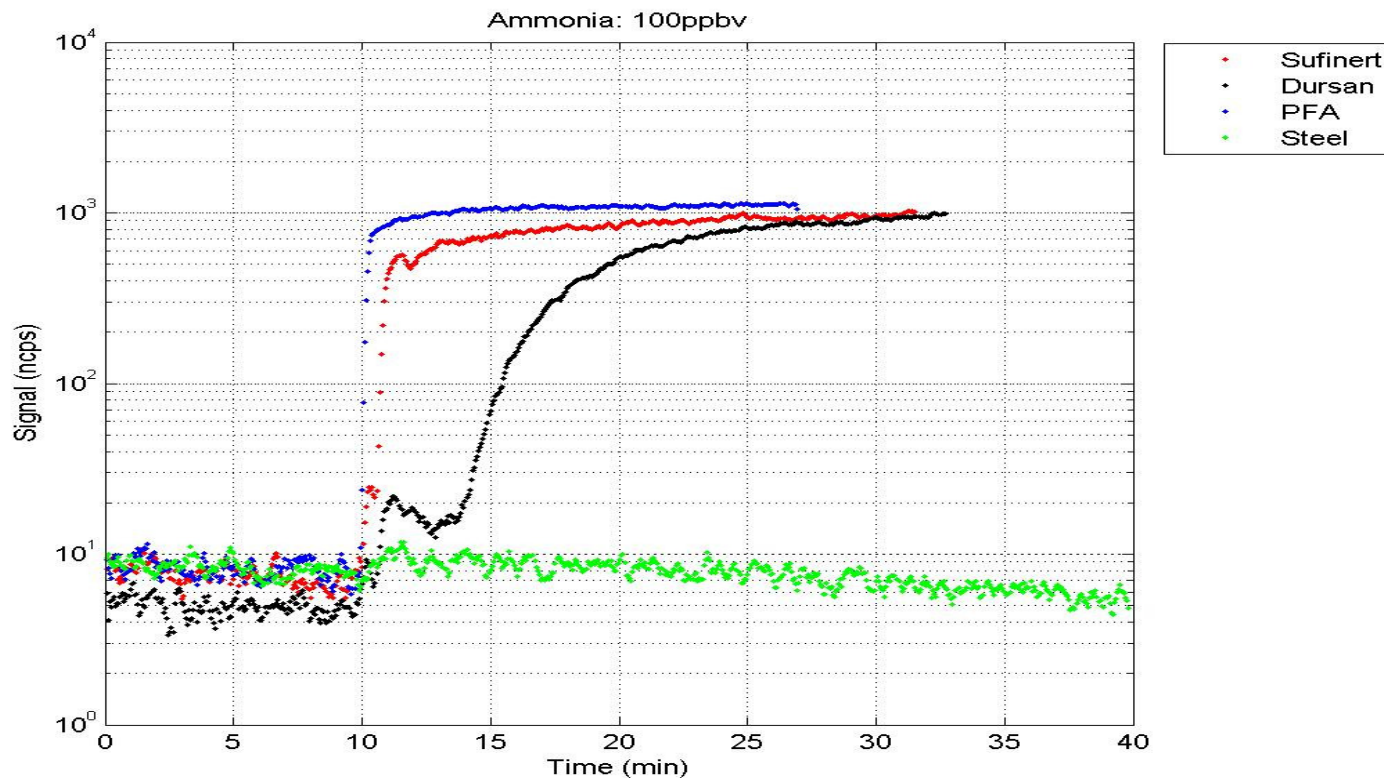


	0:00	25 hours	50 hours	75 hours
— Stainless Steel	100	19	10	4
— Dursan	100	98	96.6	98.3



# Ammonia Adsorption

100PPV, 500sccm, 1.8m tubing, min



- Measured PTR-MS signals of ammonia (m17). At  $t=10\text{min}$  the gas stream was switched in a way
- that it passed additionally the different 1.8m long lines. The PFA line seems to be best for Ammonia, while
- the steel line completely adsorbs the 100ppbv of Ammonia in the sample gas for hours. All lines were 1.8 m,
- not heated ( $30^{\circ}\text{C}$ ), sample gas flow was 500 sccm (std. ml/min) of 100 ppb of ammonia in  $\text{N}_2$ .

# Conclusion

- Consider multiple design factors to maximize system performance and reliability
  - Surface Energy (moisture resistance)
  - Surface Roughness
  - Corrosion resistance
  - Wear resistance and other physical related factors
  - Surface interaction/inertness
- Review system design, balance target system performance and environment with material capability