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THE VALUE OF THERMAL MASS²

Increase thermal comfort

- · Play as a buffer to absorb excessive solar gains
- Reduce influence from outside weather conditions changes.
- Reduce temperature fluctuations inside building

Decrease the energy bill, and the environmental footprint

- Reduce the energy consumption in summer (air conditioning)
- · Reduce the energy consumption in winter, spring, and autumn (heating)
- Contribute to system solution to avoid air conditioning.
- Release energy gained during the day, at night. (in spring, autumn, and winter)

Contribute to satisfy building regulations

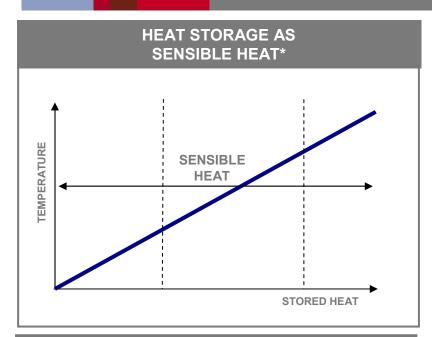
- Prevent overheating in summer
- Helps to achieve low energy rating

Helps reduce CO₂ emissions



Passive solution No maintenance costs



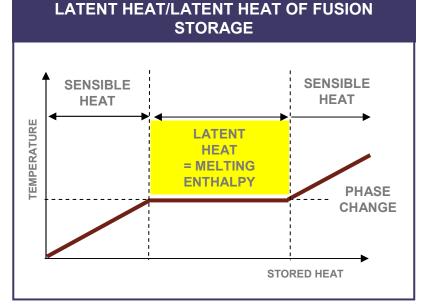


A temperature increase leads to a temperature increase of the storage medium (blue curve)

Heat capacity of the medium: Stored heat/temperature increase

Examples: STONE, BRICK, CONCRETE

HEAT STORAGE PRINCIPLE ³



The material absorbs sensible heat and starts to melt when reaching the temperature of the phase change. Until melting is completed, the material maintains a constant temperature. Once complete, any further heat transfer results in sensible heat storage. The heat stored during the melting process (melting enthalpy) is called the latent heat as the material doesn't show a temperature increase.

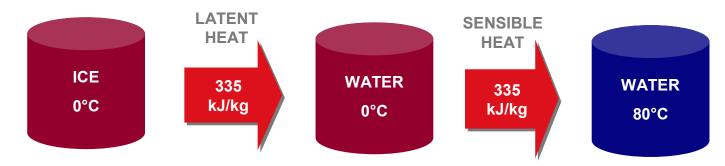
* Sensible heat is the energy exchanged by a thermodynamic system during a change in temperature. The sensible heat of a thermodynamic process may be calculated as the product of the body's mass with its specific heat and its change in temperature: $Q_{\text{Sensible}} = mc\Delta T$ where m is the mass, c is the specific heat capacity of the body at the appropriate temperature and pressure range, and ΔT is the change in temperature of the body.

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SENSIBLE vs. LATENT HEAT STORAGE 4

Water Example:



SENSIBLE vs. LATENT HEAT STORAGE

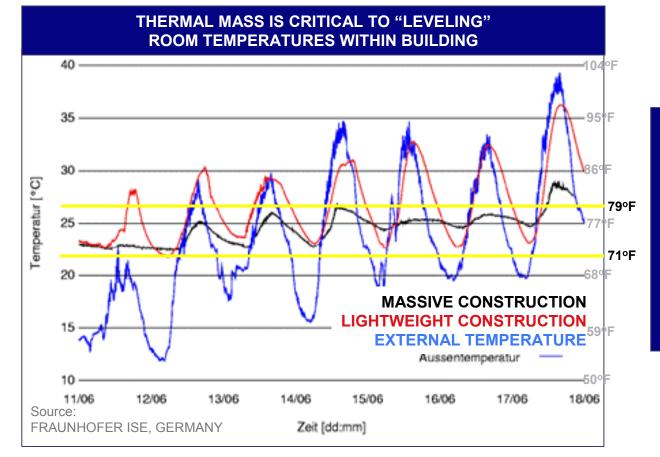
	kJ/l	kJ/kg	Comment
Sensible heat			
Granite	50	17	$\Delta T = 20^{\circ} C$
Water	84	84	$\Delta T = 20^{\circ} C$
Latent heat of melting			
Water	330	330	0 °C
Paraffin	180	200	5–130 °C
Salthydrate	300	200	5–130 °C
Salt	600-1,500	300-700	300–800 °C

Source: Harald Mehling, Luis Cabeza Bavarian Center for Applied Energy Research, ZAE Bayern, Germany

PCM CAN STORE 3-4 TIMES MORE ENERGY PER VOLUME AS IS STORED AS SENSIBLE HEAT STORAGE IN SOLIDS OR LIQUIDS FOR A TEMPERATURE INTERVAL OF 20 °C.



WHY THERMAL INERTIA IS IMPORTANT 5



THERMAL MASS EFFECT = THERMAL INERTIA

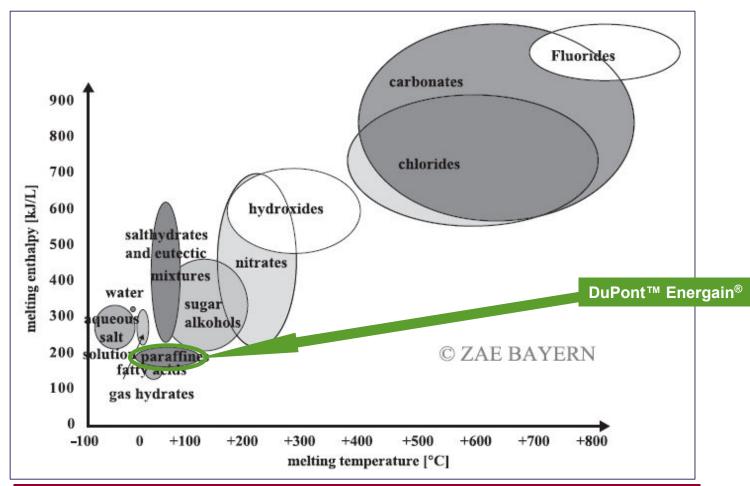
Thermal mass is the ability of a material to absorb and store heat. The thermal mass performance is determined by thermal conductivity and high specific heat capacity to maximize the heat that can be stored per kg of material.

A Lack of Thermal Mass Leads to Larger Interior Temperature Swings in

The Building Resulting in Higher Energy Demands (Especially in Summer)



WHAT IS A PHASE CHANGE MATERIAL (PCM)? 6



WHAT IS A Phase Change Material (PCM)?

PCM's such as water, paraffin, salt hydrates, *etc.* are able to absorb, store and release large amounts of thermal energy within comparatively small, defined temperature ranges by changing their physical state: *e.g.* Solid-to-liquid, liquid-to-solid, solid-to-solid or through evaporation of the storage material. The heat stored is called latent heat, therefore materials are also referred to as "LATENT HEAT STORAGE MATERIAL".

12/6/2010



DUPONT PRODUCT-TECHNOLOGY 7

DuPont[™] Energain[®]: Phase Change Material

DESIGNED TO

- Add 'effective thermal mass' to structures
- Store more energy than concrete with less mass & thickness

TECHNOLOGY

- Paraffin wax in a polymer matrix
- Dimensions: 1.0 m x 1.2 m x 5.26 mm

PERFORMANCE

- System melting point 22 °C (71°F)
- Solid at / or below ~ 18 °C (65°F)
- Once the room temperature increases, the paraffin starts liquefying and absorbs thermal energy (515 kJ/m²) at 18-24°C
- Once the room temperature drops, the paraffin starts solidifying and releases the absorbed energy back into the room
- Net effect of the absorption/de-absorption of the thermal energy is a 'leveling of the temperature swings' within the room and, hence, a lower demand on the HVAC system

SPECIFICATION

- Model simulation of building installation requirements for design:
 - Number of square meters needed ~ 0.5-1m² per m³ volume
 - Where best to install
 - Temperature reduction & energy savings estimates

VALIDATION

Real life-experiment by EDF (Electricité de France), France





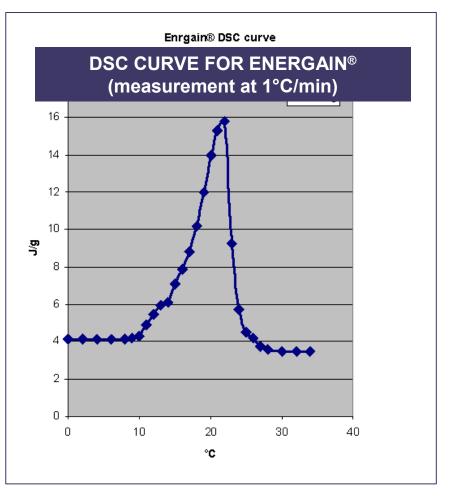
DuPont Innovation (PATENT)

ensures that the material (polymer compound / panel design) maintains its mechanical stability while the paraffin is in liquid state.



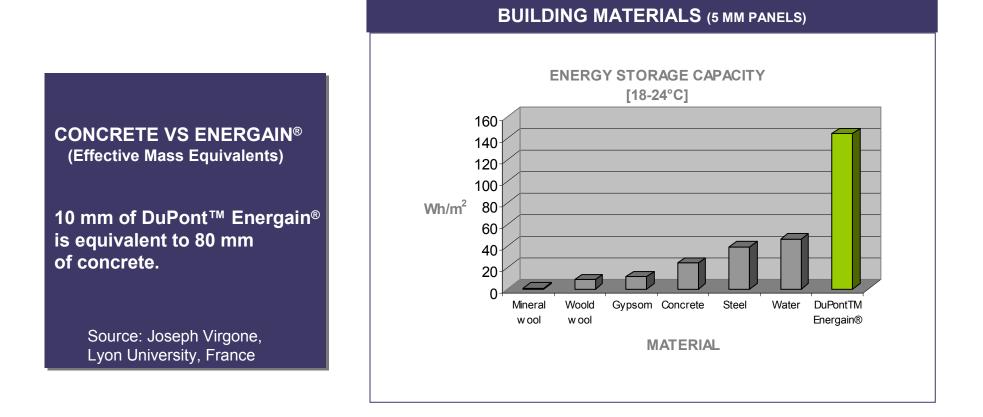
ENERGAIN[®]: ENERGY CAPACITY ⁸

ENERGAIN [®] : PROPERTY	UNIT	VALUE
Temperature window	°C	10-31
Peak temperature	°C	22
Weight	kg/m ²	4.05
Thickness	m	0.005
Vol (Mass)	kg/m ³	810
Total heat storage capacity across temperature window (10-31°C)	kJ/m ²	649
Theoretical Energy absorption	Wh/m ² .d	180



DSC: Differential scanning calorimetry is a thermoanalytical technique that measures the amount of heat required to increase the temperature of a sample for a given time step.

PCM: 'EQUIVALENT' THERMAL MASS COMPARISON

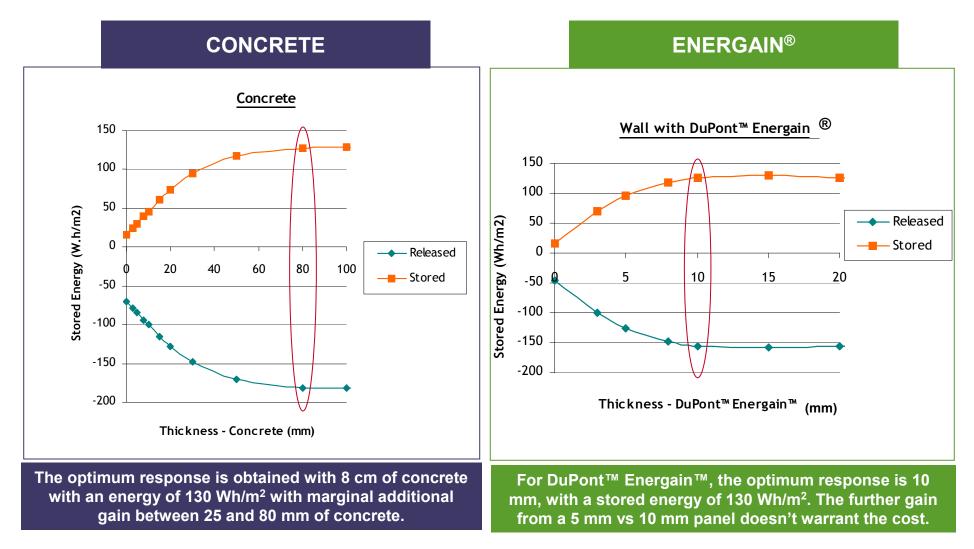


CONCRETE OFFERS ONLY 17% OF THE ENERGY STORAGE CAPACITY IN THE 18-24°C TEMPERATURE RANGE AT 5MM



9

PCM: THERMAL MASS COMPARISON



Source: Joseph Virgone, Lyon University, France



"5" EASY INSTALLATION STEPS 11



The panel can be cut to any size and fitted in place by drilling and screwing, nailing and stapling. The panels are fitted behind ordinary dry lining plasterboards.



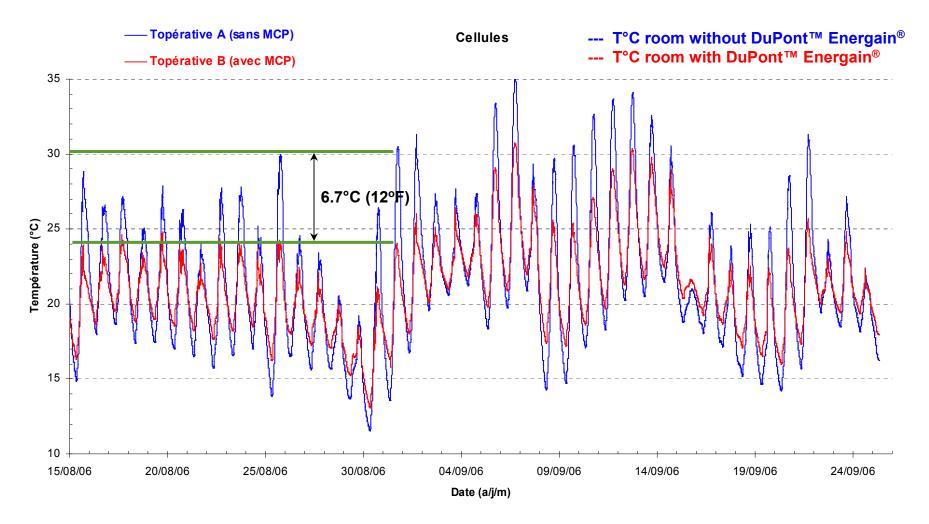
REAL-LIFE EXPERIMENT WITH EDF FRANCE¹²

OBJECTIVE: Product performance validation under seasonal impact. DuPont field test in cooperation with EDF (Electricité de France) in the Paris region.





ROOM CLIMATE MANAGEMENT 13



On Average, the PCM Room Peak Temperatures Were Reduced by 4.5°C (8°F) Relative to the Non-PCM Room During the Test Period

(Occasionally Observed Differences As High As 6.7°C)



BUILDING REFERENCES

HAMMOND HIGH SCHOOL, Swaffham, GB





Excessive internal temperature can have an adverse effect on both wellbeing and productivity – and for schoolchildren, it can prove a very unwelcome distraction.

UK Building Bulletin 101 -Ventilation of School Buildings:

1. There should be no more than 120 hours per year when the air temperature in the classroom rises above 28°C (82°F)

2. The internal air temperature when the space is occupied should not exceed 32°C (90°F)

600 m² of DuPont[™] Energain[®] installed into the ceilings of new classrooms.



BUILDING REFERENCES

NAPIER UNIVERSITY, in Edinburgh, GB 2010





Images courtesy of RMJM and ENU

650 m² of DuPont[™] Energain[®] has been installed





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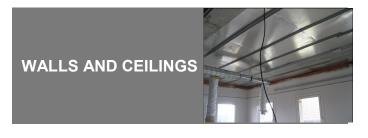
BUILDING REFERENCES PREBAT OFFICE BUILDING, Lyon, France

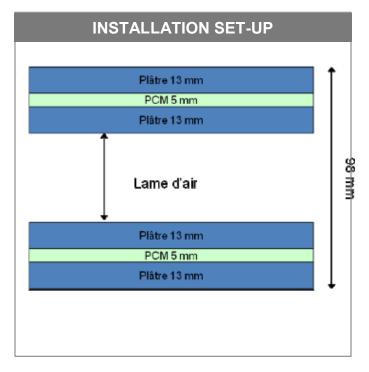


DuPont[™] Energain[®] was selected to avoid installing an air conditioning system.



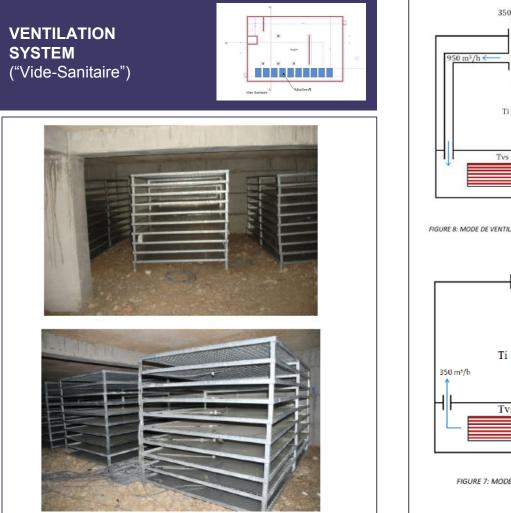
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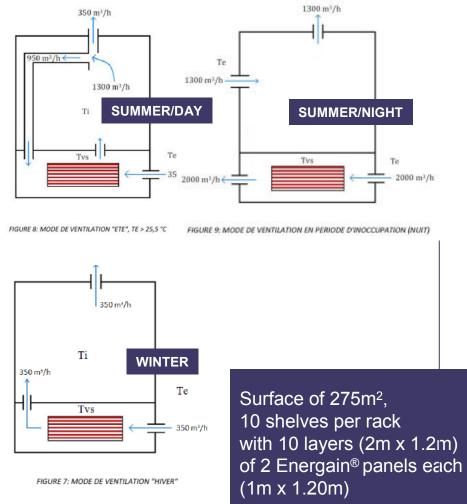






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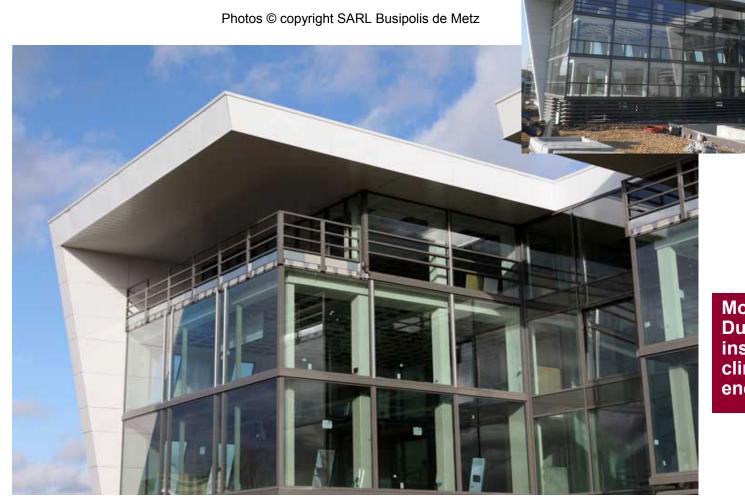




BUILDING REFERENCES BUSIPOLIS, Metz, France 2010

More than 500 m² of DuPont[™] Energain[®] installed to improve room climate and reduce energy costs.

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BUILDING REFERENCES **ZERO CARBON HOUSE, Kent, GB 2009**



CROSSWAY HOUSE BY RICHARD HAWKES

Finalist in Grand Designs Awards 2009.

Energain® installed to add thermal mass.







BUILDING REFERENCES PRIVATE VILLA, PRAGUE, CZECH REPUBLIC 2010





"Manipulation and installation of the DuPont™ Energain[®] panels were quick and easy, a perfect solution for this project", said Mr. Rychlý, sales manager of Kerilit (supplied and installed).





The miracles of science™

